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(1*E*,2*E*)-1,2-Bis[(*E*)-3-phenylallylidene]-hydrazine

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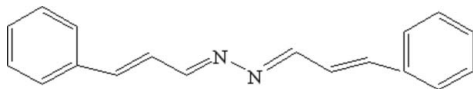
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Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.041; wR factor = 0.094; data-to-parameter ratio = 15.2.

The title compound [alternatively called (*E*)-3-phenylprop-2-enal azine], $\text{C}_{18}\text{H}_{16}\text{N}_2$, was synthesized by the reaction of hydrocinnamaldehyde with hydrazine hydrate. The nearly planar molecule is centrosymmetric, with the mid-point of the N—N bond lying at an inversion center.

Related literature

For related literature, see: Jiang & Hu (2004); Kundu *et al.* (2005); Trujillo *et al.* (1997); Zheng & Zhao (2006); Zhao *et al.* (2006).



Experimental

Crystal data

$\text{C}_{18}\text{H}_{16}\text{N}_2$	$V = 738.0$ (13) Å ³
$M_r = 260.33$	$Z = 2$
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation
$a = 17.11$ (2) Å	$\mu = 0.07$ mm ⁻¹
$b = 5.456$ (5) Å	$T = 293$ (2) K
$c = 7.988$ (7) Å	$0.55 \times 0.21 \times 0.12$ mm
$\beta = 98.25$ (4)°	

Data collection

Rigaku R-AXIS RAPID IP
diffractometer
Absorption correction: none
5393 measured reflections

1381 independent reflections
844 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.034$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.041$
 $wR(F^2) = 0.094$
 $S = 0.95$
1381 reflections

91 parameters
H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.09$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.14$ e Å⁻³

Data collection: *TEXRAY* (Molecular Structure Corporation, 1999); cell refinement: *TEXRAY*; data reduction: *TEXSAN* (Molecular Structure Corporation, 1999); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEX* (McArdle, 1995); software used to prepare material for publication: *SHELXL97*.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: XU2335).

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supplementary materials

Acta Cryst. (2007). E63, o4443 [doi:10.1107/S1600536807051707]

(1*E*,2*E*)-1,2-Bis[(*E*)-3-phenylallylidene]hydrazine

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Comment

The azine derivatives have been investigated in terms of their coordination chemistry and their special properties (Zheng & Zhao, 2006; Kundu *et al.*, 2005; Jiang & Hu, 2004; Trujillo *et al.*, 1997). The synthesis and crystal structure of the title compound (Fig. 1) is reported here.

In the molecule of title molecule, there is an inversion center at the midpoint of the N1—N1A single bond [symmetry code A, $-x, -y, -z + 1$], and all of the atoms are approximately coplanar. The single N—N bond length is 1.398 (3) Å (Table 1), which is consistent with those found in related azine compounds (Zheng & Zhao, 2006; Zhao *et al.*, 2006). The bond angle of N(1)—C(9)—C(8) is 120.9 (2)°, showing the atom C9 is of sp^2 hybridization.

Experimental

The title compound was prepared by the reaction of hydrocinnamaldehyde and hydrazine hydrate with a ratio of 2:1 under reflux in the mixture of ethanol and DMF for 2 h. After one week, yellow prismatic crystals were obtained by slow evaporation at room temperature.

Refinement

All H atoms were placed in calculated positions and treated using a riding model with C—H = 0.93 Å and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$.

Figures

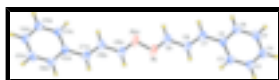


Fig. 1. View of the molecular structure with displacement ellipsoids drawn at the 30% probability level for non-H atoms. [Symmetry code: (A) $-x, -y, -z + 1$.]

(1*E*,2*E*)-1,2-Bis[(*E*)-3-phenylallylidene]hydrazine

Crystal data

C₁₈H₁₆N₂

$M_r = 260.33$

Monoclinic, $P2_1/c$

Hall symbol: $-P 2ybc$

$a = 17.11$ (2) Å

$b = 5.456$ (5) Å

$F_{000} = 276$

$D_x = 1.172$ Mg m⁻³

Mo $K\alpha$ radiation

$\lambda = 0.71073$ Å

Cell parameters from 1381 reflections

$\theta = 3.6$ – 26.0°

$\mu = 0.07$ mm⁻¹

supplementary materials

$c = 7.988$ (7) Å
 $\beta = 98.25$ (4)°
 $V = 738.0$ (13) Å³
 $Z = 2$

$T = 293$ (2) K
Prism, yellow
 $0.55 \times 0.21 \times 0.12$ mm

Data collection

Rigaku R-Axis RAPID IP diffractometer
Radiation source: fine-focus sealed tube
Monochromator: graphite
 $T = 293$ (2) K
 ω scans
Absorption correction: none
5393 measured reflections
1381 independent reflections

844 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.034$
 $\theta_{\text{max}} = 26.0^\circ$
 $\theta_{\text{min}} = 3.6^\circ$
 $h = -21 \rightarrow 21$
 $k = -6 \rightarrow 6$
 $l = -9 \rightarrow 9$

Refinement

Refinement on F^2
Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.041$
 $wR(F^2) = 0.094$
 $S = 0.95$
1381 reflections
91 parameters
Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map
Hydrogen site location: inferred from neighbouring sites
H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0491P)^2]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.09 \text{ e \AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.14 \text{ e \AA}^{-3}$
Extinction correction: none

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
N1	0.04070 (6)	-0.0045 (3)	0.49829 (15)	0.0722 (4)
C1	0.26140 (7)	0.4619 (2)	0.34602 (14)	0.0481 (3)

C2	0.32091 (8)	0.3004 (2)	0.41067 (17)	0.0517 (4)
H2A	0.3085	0.1627	0.4701	0.062*
C3	0.39765 (8)	0.3412 (3)	0.38813 (17)	0.0568 (4)
H3A	0.4367	0.2307	0.4321	0.068*
C4	0.41760 (9)	0.5435 (3)	0.30128 (17)	0.0608 (4)
H4A	0.4698	0.5701	0.2863	0.073*
C5	0.35992 (9)	0.7059 (3)	0.23671 (19)	0.0626 (4)
H5A	0.3731	0.8428	0.1773	0.075*
C6	0.28275 (8)	0.6675 (3)	0.25939 (16)	0.0562 (4)
H6A	0.2443	0.7802	0.2164	0.067*
C7	0.17886 (8)	0.4207 (3)	0.36486 (16)	0.0579 (4)
H7A	0.1439	0.5477	0.3309	0.070*
C8	0.14843 (8)	0.2210 (3)	0.42545 (18)	0.0609 (4)
H8A	0.1825	0.0927	0.4619	0.073*
C9	0.06635 (8)	0.1918 (3)	0.43782 (18)	0.0657 (4)
H9A	0.0312	0.3173	0.4014	0.079*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
N1	0.0402 (6)	0.0995 (10)	0.0762 (8)	-0.0123 (7)	0.0061 (6)	0.0094 (8)
C1	0.0463 (8)	0.0529 (8)	0.0449 (7)	-0.0053 (6)	0.0062 (5)	-0.0037 (7)
C2	0.0490 (8)	0.0503 (8)	0.0561 (8)	-0.0049 (6)	0.0086 (6)	0.0018 (7)
C3	0.0481 (8)	0.0587 (9)	0.0638 (9)	0.0000 (7)	0.0090 (6)	-0.0045 (7)
C4	0.0531 (9)	0.0677 (11)	0.0644 (9)	-0.0145 (8)	0.0178 (7)	-0.0093 (8)
C5	0.0723 (11)	0.0556 (9)	0.0622 (9)	-0.0185 (8)	0.0178 (8)	0.0014 (7)
C6	0.0616 (9)	0.0504 (9)	0.0561 (8)	-0.0006 (7)	0.0064 (7)	0.0009 (7)
C7	0.0467 (8)	0.0688 (10)	0.0573 (8)	0.0001 (7)	0.0039 (6)	0.0010 (8)
C8	0.0433 (9)	0.0803 (11)	0.0588 (9)	-0.0071 (7)	0.0059 (6)	0.0042 (8)
C9	0.0456 (9)	0.0900 (12)	0.0606 (9)	-0.0084 (8)	0.0042 (6)	0.0028 (9)

Geometric parameters (\AA , $^\circ$)

N1—C9	1.278 (2)	C4—H4A	0.9300
N1—N1 ⁱ	1.398 (3)	C5—C6	1.374 (2)
C1—C2	1.389 (2)	C5—H5A	0.9300
C1—C6	1.3944 (19)	C6—H6A	0.9300
C1—C7	1.459 (2)	C7—C8	1.329 (2)
C2—C3	1.369 (2)	C7—H7A	0.9300
C2—H2A	0.9300	C8—C9	1.431 (2)
C3—C4	1.373 (2)	C8—H8A	0.9300
C3—H3A	0.9300	C9—H9A	0.9300
C4—C5	1.371 (2)		
C9—N1—N1 ⁱ	112.04 (17)	C4—C5—H5A	119.8
C2—C1—C6	117.70 (14)	C6—C5—H5A	119.8
C2—C1—C7	122.21 (14)	C5—C6—C1	120.87 (14)
C6—C1—C7	120.09 (13)	C5—C6—H6A	119.6
C3—C2—C1	120.92 (14)	C1—C6—H6A	119.6

supplementary materials

C3—C2—H2A	119.5	C8—C7—C1	126.82 (14)
C1—C2—H2A	119.5	C8—C7—H7A	116.6
C2—C3—C4	120.66 (14)	C1—C7—H7A	116.6
C2—C3—H3A	119.7	C7—C8—C9	123.78 (15)
C4—C3—H3A	119.7	C7—C8—H8A	118.1
C5—C4—C3	119.49 (15)	C9—C8—H8A	118.1
C5—C4—H4A	120.3	N1—C9—C8	120.86 (16)
C3—C4—H4A	120.3	N1—C9—H9A	119.6
C4—C5—C6	120.34 (15)	C8—C9—H9A	119.6
C6—C1—C2—C3	-0.73 (18)	C7—C1—C6—C5	-178.33 (12)
C7—C1—C2—C3	178.68 (12)	C2—C1—C7—C8	-8.1 (2)
C1—C2—C3—C4	0.2 (2)	C6—C1—C7—C8	171.33 (13)
C2—C3—C4—C5	0.0 (2)	C1—C7—C8—C9	-178.97 (12)
C3—C4—C5—C6	0.3 (2)	N1 ⁱ —N1—C9—C8	179.17 (13)
C4—C5—C6—C1	-0.9 (2)	C7—C8—C9—N1	-179.57 (14)
C2—C1—C6—C5	1.08 (19)		

Symmetry codes: (i) $-x, -y, -z+1$.

Fig. 1

